



Bismuth, a brittle, silvery-white, less toxic metal placed after the transition elements in the periodic table. It has a low melting point and thermal conductivity. It gets easily oxidized in air. It is naturally available in the form of bismuth oxides and bismuth sulfides with outstanding electronic and optical properties. Owing to its excellent redox reactions, various bismuth compounds such as bismuth oxides, sulfides, selenides, and oxyhalides were investigated recently in the supercapacitor technologies. In recent days, various efforts have been taken by various researcher groups to develop bismuth-based composites with high SCP, excellent cyclic stability respectively. Very few researches constructed the supercapacitor devices in terms of household appliances.

## Stock no:

## NS6130-12-001588

## Chemical Identifiers

Chemical name (Bi2O3)(CoO)(ZnO)

APS 80-100nm Density 5.6g/cm3

## **Applications**

- Pharmaceutical, cosmetic, and chemical products
- Medicines, especially for treating infection problems in the stomach (peptobismol) and eyes (eye drops). The active component in peptobismol is bismuth oxide salicylate.
- Cosmetics, as bismuth oxychloride to give a shine to lipsticks, nail varnishes, and hair lacquers.



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